

Ring transformations of 2,3-dihydroisoxazoles via azomethine ylides—formation of annulated 5- and 7-membered N-heterocycles[☆]

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Abstract—On thermal activation the 2,3-dihydroisoxazoles **12–14** are transformed into annulated dihydroazepines **15–17** as main products, besides minor amounts of the corresponding pyrrole derivatives **18–20**. In the proposed mechanism the azomethine ylides of type **III** and **VI** are involved as intermediates which undergo 1,5- and 1,7-ring closure reactions, respectively. © 2001 Elsevier Science Ltd. All rights reserved.

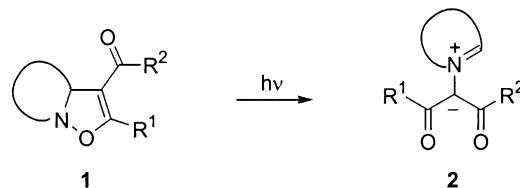
1. Introduction

Among the various reduced isoxazole derivatives, the 2,3-dihydroisoxazoles belong to the less common isomers because of their notorious chemical instability under many conditions.² Besides decomposition material, the mainly isolated products are, if at all, pyrroles, 4-oxazolines, acyl aziridines and acyclic enamines, respectively, the kind and ratio of the products depending strongly on the substitution pattern.³ However, as a common feature of many of these reactions the transformations are initiated by NO-cleavage and subsequent rebonding affording acyl aziridines which react further by ring opening to give azomethine ylides, the supposedly precursors of the observed variety of products (Scheme 1).^{2,3}

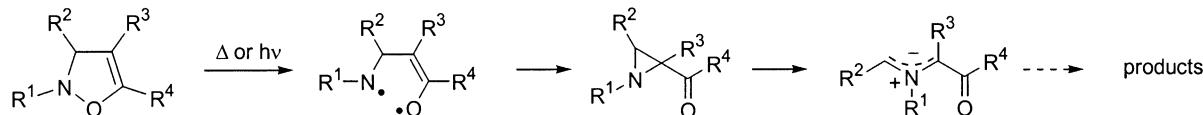
In earlier work we have shown that on photochemical activation of acyl-isoxazolines of type **1** the corresponding doubly stabilized azomethine ylides **2** can be isolated, and

also evidence for the appearance of aziridines have been obtained (Scheme 2).⁴

The transient formation of azomethine ylides during reactions of those heterocycles prompted us to evaluate the potential of 2,3-dihydroisoxazoles as precursor of azomethine ylides. In particular, investigations were performed that should allow 1,5- and/or 1,7-dipolar cyclization reactions of the intermediate conjugated dipole species as deactivation pathways.⁵



Scheme 2.

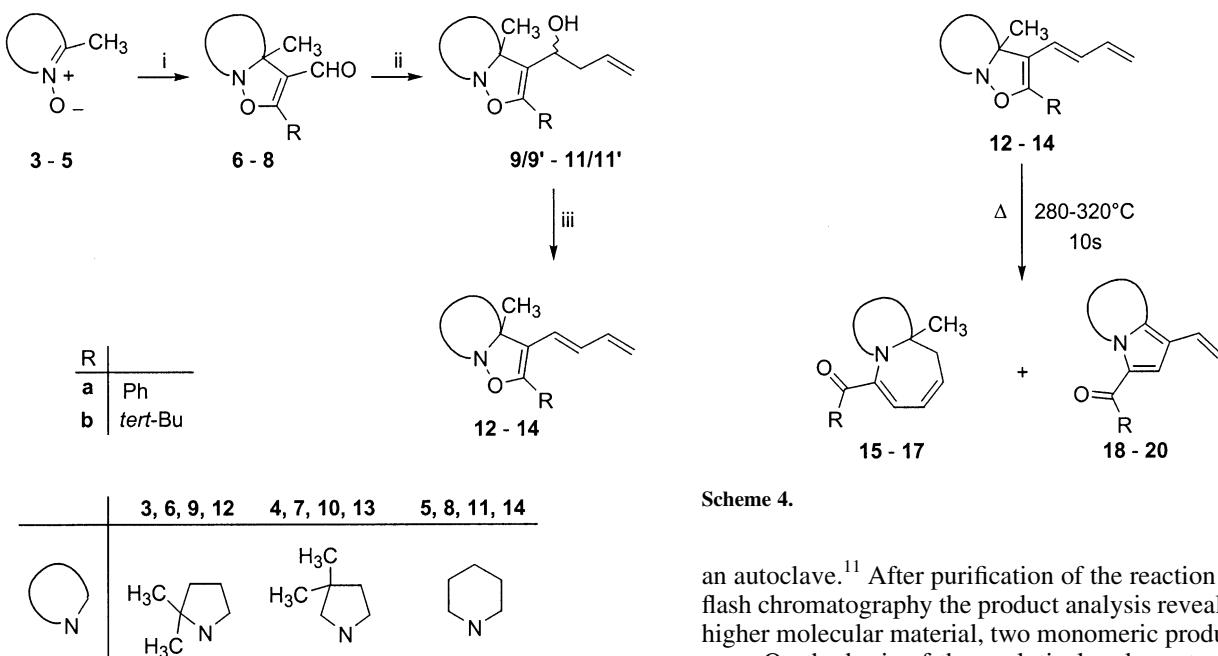


Scheme 1.

* Part 5 in the series ‘The pyrido[1,2-*a*]azepine system’, for Part 4 see Ref. 1.

Keywords: 2,3-dihydroisoxazoles; azepines; azomethine ylides; dipolar cyclizations.

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Scheme 4.

an autoclave.¹¹ After purification of the reaction mixture by flash chromatography the product analysis revealed, besides higher molecular material, two monomeric products in each case. On the basis of the analytical and spectroscopic data the structure of the main products was identified as the isomeric bicyclic dihydroazepine systems **15–17**, whereas the minor components turned out to be the 1,2-annulated pyrroles **18–20**, which differ from the starting isoxazolines in the formal loss of a CH₄ unit; the yields are given in Table 1. For the structure elucidation of the reaction products special NMR techniques have been applied including HMBC spectra.¹²

2. Results and discussion

Here we present results of our studies with 2,3-dihydroisoxazoles bearing a butadienyl group at C-3 (Scheme 3). The heterocyclic ring was constructed by 1,3-dipolar cycloaddition of the known cyclic nitrones **3–5** with phenyl and *tert*-butyl propionic aldehyde, respectively. As predicted by FMO theory and found experimentally in other cases,^{6,7} only the shown regioisomers **6–8** were formed. The further functionalization included Grignard reactions with allyl magnesium bromide affording the secondary alcohols **9–11** as diastereomeric mixtures⁸ which were finally transformed into the isoxazolines **12a,b–14a,b** by water elimination.⁹ The yields for most steps were between 70 and 90%, with the exception of **6b** (55%), **12b** (64%), **14a** (58%) and **14b** (40%).

The thermal reactions were performed with ca. 10⁻² molar solutions of **12a,b–14a,b** in benzene under short-time thermolysis conditions¹⁰ at 280–320°C with a contact time of ca. 10 s. This technique was superior to other methods such as flash-vacuum thermolysis or heating of solutions in

For the transformation of **12–14** into the reaction products **15–17** and **18–20** a multi-step pathway has to be envisaged (Scheme 5). The first part of the mechanistic interpretation is based on the general reactivity of 2,3-dihydroisoxazoles² and includes the ring transformation of the compounds of general type **I** into the bicyclic aziridines **II** and subsequent C–C bond cleavage with formation of the azomethine ylide intermediates **III**. The conjugated dipole system then undergoes a 6π ring closure¹³ affording the pyrrolines **IV** which under the harsh reaction conditions suffer CH₃-H elimination¹⁴ to give the observed pyrroles **VII**. For rationalizing the formation of the main products **V** a preceding isomerization of the central double bond of the conjugated dipolar system must take place (**III**→**VI**). Analogous (E)↔(Z) isomerizations have been observed with diazo derivatives,¹⁵ carbonyl ylides¹⁶ and nitrones.¹⁷ In case of a deprotonated 2-azadiaryl-heptatriene the geometrical change takes place even at -20°C.¹⁸ Finally, the *syn*-arranged species **VI** undergoes a 1,7-dipolar electrocyclization process affording the azepine derivatives **V**.^{1,19,20}

Table 1. Thermolysis products of **12a,b**, **13a,b** and **14a,b**

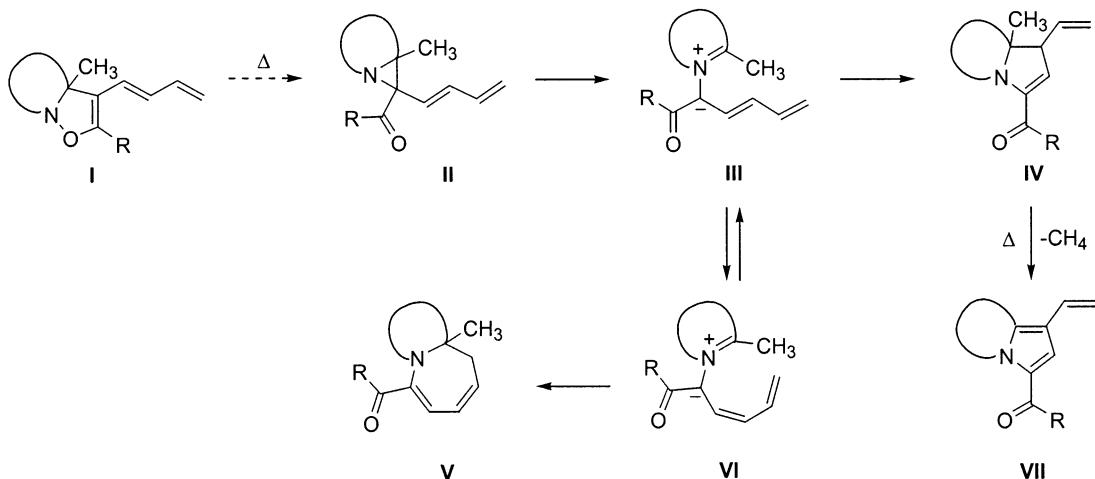
	15	18	16	19	17	20
a	59 ^a	9 ^a	48 ^a	3 ^a	42 ^b	7 ^b
b	13 ^b	–	38 ^b	tr ^b	41 ^b	tr ^b

Yield in %, after chromatography.

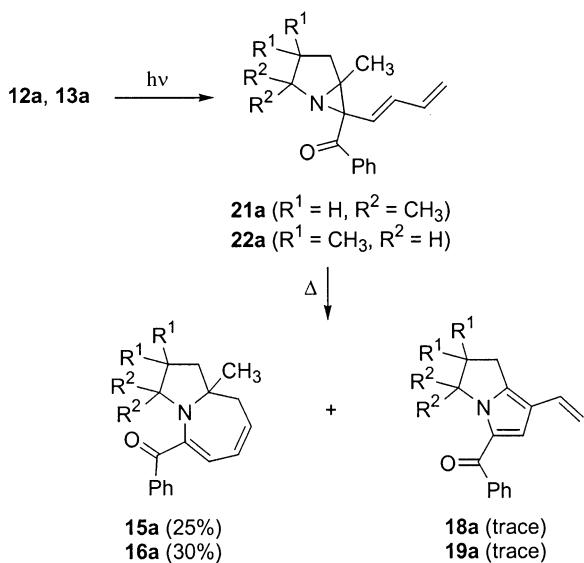
^a 320°C/10 s.

^b 280°C/10 s.

Further evidence for the occurrence of intermediates of type **II** was obtained by independent thermolysis experiments with the bicyclic aziridines **21a/22a** which can be selectively obtained on photoinduced transformation of the annulated isoxazolines **12a/13a** (Scheme 6): on short-time thermolysis of **21a** (320°C) and **22a** (280°C) the same ring expansion products **15/18** and **16/19**, respectively, are formed²¹ as have been observed with the 2,3-dihydroisoxazoles **12a** and **13a** as starting materials in this work.



Scheme 5.



Scheme 6.

3. Conclusion

In summary, the experiments with various 4-isoxazolines have shown that on thermal activation the main reaction pathway affords azomethine ylides as key intermediates which in the presence of appropriate π -substituents like a butadienyl group undergo electrocyclic ring closure reactions to annulated pyrrole and azepine derivatives, respectively. An extension of the work with 4-isoxazolines bearing different side chains as well as more studies on the photochemical reactivity of such heterocycles are in progress.

4. Experimental

4.1. General

Melting points are uncorrected. ^1H NMR: 250, 400 and 500 MHz; ^{13}C NMR: 100 or 125 MHz, Brucker WM 250, WM 400, WM 500; CDCl_3 as solvent and TMS as internal

standard; for assignments ATP, COSY, HMBC, NOESY techniques were used. IR: Perkin Elmer PE 297 Infracord FT-IR. UV: Perkin-Elmer Lambda 15. MS: Finnigan MAT 44 S spectrometer with Datasystem MAT SS 200 using electron impact ionization (EI, 70 eV) or chemical ionization (CI, 170 eV) with isobutane. Elemental analyses: Perkin-Elmer Elemental Analyzer 240. All reactions were carried out in flame-dried glassware under N_2 atmosphere unless otherwise stated. Analytical TLC: precoated silica gel Merck 60 F-254 (0.2 mm). Flash chromatography: ICN-Biomedicals silica gel (ICN Silica 32-36). Short-time thermolysis apparatus: a vertical, externally heated Pyrex tube (37×3 cm) packed with Raschig rings (Pyrex, 4×4 mm); packed height 18 cm, heating zone 30 cm; addition of the solutions through a dosing funnel (Normag N 8056) in a N_2 -steam (flow rate 0.75 l/h), dropping rate 13 ml/h; temperature $\pm 10^\circ\text{C}$, contact time ca. 10 s.^{10,11}

4.2. Procedure for the preparation of the annulated dihydroisoxazoles 6–8

6a,b: To a solution of the nitrone **3**²² (ca. 4 mmol) in 5–10 ml dry CH_2Cl_2 of diethyl ether was added under argon 3-phenylpropynal²³ and 3-*tert*-butylpropynal,^{17b} respectively, at r.t. The reaction mixture was further stirred under argon and light exclusion at room temperature for 24 h. After evaporation of the solvent the oily residue was purified by flash chromatography (SiO_2 , cyclohexane/ethyl acetate 20:1). The compounds **7a,b** and **8a,b** were likewise prepared from the nitrones **4**²⁴ and **5**,²⁵ respectively, following the known protocol.^{4b}

4.2.1. 3a,6,6-Trimethyl-2-phenyl-3a,4,5,6-tetrahydro-pyrrolo[1,2-*b*]isoxazol-3-carbaldehyde (6a). The reaction of **3** (440 mg, 3.46 mmol) with 3-phenylpropynal (455 mg, 3.5 mmol) in 5 ml dry CH_2Cl_2 gave 717 mg (80%) of **6a** as pale yellow oil. ^1H NMR (250 MHz): $\delta = 1.27$ (s, 3 H, CH_3), 1.34 (s, 3 H, CH_3), 1.67 (s, 3 H, CH_3), 1.70–1.85 (m, 2 H, CH_2), 2.02 (ddd, 1 H, CH_2 , $^2J=13.1$ Hz, $^3J=9.1$ Hz, $^3J=6.7$ Hz), 2.43 (ddd, 1 H, CH_2 , $^2J=13.1$ Hz, $^3J=7.0$ Hz, $^3J=4.9$ Hz), 7.43–7.62 (m, 5 H, PhH), 9.64 (s, 1 H, CHO). ^{13}C NMR (100 MHz): $\delta = 22.9$ (CH_3), 27.2 (CH_3), 28.2 (CH_3), 36.2 (CH_2), 36.5 (CH_2), 69.8 (C_q), 77.2 (NC_q), 119.4 ($=\text{C}_q$),

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